

Impact of Nitrogen amendments on Soil Enzyme Dynamics under Simulated Wetland Ecosystem

ABSTRACT

Aims: To evaluate the influence of nitrogen amendments on soil enzyme dynamics in a long term incubation experiment.

Study Design: An *in vitro* simulated wetland ecosystem designed with rhizosphere soil was enriched with different N sources.

Place and Duration of Study: The study was conducted at Biocatalysts Laboratory, Tamil Nadu Agricultural University, Coimbatore, India. An incubation experiment ran for 150 days, to determine the temporal changes of soil enzyme activities.

Methodology: There were five treatments replicated thrice. The N enrichment included in the treatments were aerated except S1 as detailed below: rhizosphere soil (S1), rhizosphere soil without enrichment (S2), combined NH_4Cl and KNO_2 enriched rhizosphere soil (S3), KNO_2 enriched rhizosphere soil (S4) and NH_4Cl enriched rhizosphere soil (S5).

Results: The soil enzymes such as dehydrogenase ($24.59 \mu\text{g TPF g}^{-1} \text{soil day}^{-1}$), urease ($49.27 \mu\text{g NH}_3 \text{g}^{-1} \text{soil}$) and acid phosphatase ($38.57 \mu\text{g PNP g}^{-1} \text{soil h}^{-1}$) were observed maximum in NH_4Cl enriched rhizosphere soil (S5) on 70 DAI (days after incubation). While, highest alkaline phosphatase ($53.40 \mu\text{g PNP g}^{-1} \text{soil h}^{-1}$) and fluorescein diacetate ($7.57 \mu\text{g fluorescein g}^{-1} \text{soil h}^{-1}$) was registered on 70 DAI in KNO_2 enriched soil (S4) and $\text{KNO}_2 + \text{NH}_4\text{Cl}$ (S3), respectively. However, all the enzyme activities, irrespective of treatments, showed an increasing trend up to 70 DAI and thereafter, declined gradually.

Conclusion: Enzyme activities registered maximum in NH_4Cl enriched rhizosphere soil (S5) than other enrichments. Basal N application as ammoniacal form (NH_4^+) triggers efficient trade-offs between soil functions in the wetland ecosystem whereas, combined sources contribute to microbial biomass and redox status of the soil.

Keywords: Simulated wetland ecosystem; Nitrogen enrichment; Incubation; Soil enzymes, Ammoniacal nitrogen

1. INTRODUCTION

Wetlands are the unique, productive ecosystem that serves as carbon sinks, source, and transformers of nutrients [18]. Nitrogen is arguably a crucial nutrient in relating primary productivity and species diversity in the wetland ecosystem [31]. Imposing climate change *ie.*,

17 increased temperature and CO₂ in wetland, increase N mineralization, and microbial activities,
18 respectively. Hence the function of wetland purely relies on the extensive interaction between
19 water and wetland soil and thereby enhances the function of soil enzymes [21].

20

21 Soil enzymes maintain soil health and pave the way for the sustainable agricultural ecosystem.
22 The enzymatic activity in the soil is contributed primarily from microbial resources, intracellular,
23 extracellular and cell-associated enzymes, which are directly proportional to soil microbial
24 biomass [14]. These soil enzyme activities may serve as biological indicators and actively
25 change within the plant-soil system. Moreover, soil enzymes are closely linked to nutrient
26 cycling and act as buffers in mediating the soil functions. Therefore, soil enzymes integrate
27 information on both the microbial status and the physicochemical conditions of soil, showing a
28 rapid response to any changes in soil management practices [12]. Soil health was predicted
29 based on the key activities of the extracellular enzymes such as dehydrogenase, phosphatase,
30 urease and fluorescein diacetate in the soil profiles [16].

31 Soil dehydrogenase is an extracellular enzyme that occurs in all viable microbial cells and
32 thereby reflects the total oxidative activity of microbial biomass. Dehydrogenase usually exists
33 as an integral part of intact cells [11] and also sturdily related to soil organic matter and N cycle
34 [4]. Similarly, Urease activity in soil is an important index to evaluate soil organic matter and N
35 status of the soil. Application of NO₃⁻-N and NH₄⁺-N steadily influence soil urease activities [8].

36 On the contrary, phosphatase is a critical player in P mineralization [10] that exists in two forms:
37 Phosphodiesterases (PDE) and Phosphomonoesterases (PME). Soil generally contains large
38 quantities of intracellular and extracellular phosphatases, and the addition of glucose and
39 inorganic NH₄Cl to the soil stimulates PME at pH 6.5 and thereby makes it an available form to
40 the plants. As the microbial biomass reaches its peak, phosphatase activities tend to increase
41 rapidly. However, a prolonged period of incubation time has a negative impact on phosphatase
42 activities [13]. Fluorescein diacetate (FDA) assay is a marker to assess the total microbial
43 function in the soil. FDA undergoes hydrolysis by esterases, proteases and lipases, the
44 enzymes responsible for microbial decomposition of organic matter in the soil [1].

45 The N amendments are considered as a strategy to hasten soil microbial process and stimulate
46 associated wetland functions. Organic amendments such as compost, straw, and topsoil have
47 been shown to increase soil C and N pools [3]. Furthermore, while organic amendments
48 stimulate a balance in soil structure-functional relationships, it is unknown whether inorganic
49 amendments also impact specific nutrient geocycles with the highest lability. The present study
50 was aimed to evaluate the influence of nitrogen amendments on soil enzyme dynamics in a
51 long term incubation experiment.

52 2. MATERIALS AND METHODS

53 2.1 Sample collection for simulated wetland ecosystem

54 Soil samples were collected from the rice field, Wetland, Tamil Nadu Agricultural University,
55 Coimbatore (11.0160° N and 76.9703° E). Soil samples (0-20 cm) in triplicates collected from
56 the rice rhizosphere region were placed in sterile plastic bags, sealed, and transported to the
57 laboratory with ice. Plant residues, root samples, and stones were removed before each
58 replicate of a sample was homogenized. A simulated wetland ecosystem was set up, to clearly
59 envisage the influence of simulated environment on the nitrifiers at *in vitro* condition.

60 **2.2 Experimental design**

61 Glass containers filled with 5 kg of homogenized soil sample were exposed to the flooded
62 conditions as that of the rice field by saturating the soil with two litres of distilled water.
63 Subsequently, the set up was aerated through an airlifting motor pump with constant pressure
64 to favour the growth of both aerobic and facultative microorganisms in the soil. The rhizosphere
65 soil in glass containers was amended with 0.5% inorganic N sources such as NH_4Cl and KNO_2 .
66 The treatment and enrichment details are as below:

67 ***Rhizosphere soil alone (S1)***

68 ***Aerated rhizosphere soil (S2)***

69 ***Aerated rhizosphere soil amended with NH_4Cl + KNO_2 (S3)***

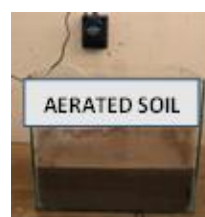
70 ***Aerated rhizosphere soil amended with KNO_2 (S4)***

71 ***Aerated rhizosphere soil amended with NH_4Cl (S5)***

72 The experimental set up of the simulated wetland ecosystem was depicted in Fig. 1. The
73 simulated wetland system was incubated for 150 d at room temperature to study the temporal
74 changes in soil enzymatic activities. Sampling was done at different intervals viz., 0, 35, 70 and
75 135 DAI (days after incubation). The reason behind the sampling days up to 135 days is to
76 facilitate the microbial build-up in the soil. At each sampling intervals, the sample was collected
77 at different points in the glass container, pooled and then analyzed by the quadrant method of
78 sample collection.



S1



S2



S3



S4



S5

Fig. 1. Experimental Set up (Simulated Wetland Ecosystem)

S1 - Rhizosphere soil; S2 - Rhizosphere soil with aeration; S3 - combined NH_4Cl and KNO_2 enriched rhizosphere soil with aeration; S4 - KNO_2 enriched rhizosphere soil with aeration; S5 - NH_4Cl enriched rhizosphere soil with aeration.

79 **2.3 Temporal dynamics of soil enzymes**

80 **2.3.1 Dehydrogenase (DHA)**

81 The dehydrogenase activity was determined spectrophotometrically at 485 nm by measuring
82 triphenyl tetrazolium formazan released from 5 g of soil after 24 h of incubation at 37°C [9]. It is
83 expressed as μg of TPF released g^{-1} soil hour⁻¹.

84 **2.3.2 Urease (URE)**

85 Urease activity was measured colourimetrically with 5 g of soil added with 0.2 mL of toluene
86 and 9 mL of Tris-hydroxymethylaminomethane (THAM) buffer (0.05 M, pH 9.0) and incubated
87 for 2 h at 37°C, according to the method of [5]. The urease activity was expressed in μg of NH_3
88 released g^{-1} soil h⁻¹.

89 **2.3.3 Phosphatase**

90 Acid phosphatase (ACP) was measured with the addition of 0.2 mL of toluene and 4 mL of
91 modified universal buffer (pH 6.5) and followed by 1 mL of 0.05M *p*-nitrophenyl phosphate (pH
92 6.5) to 1 g of soil and kept for 1 h incubation. After 1 h, 1 mL of 0.5 M calcium chloride and 4 mL
93 of 0.5 M NaOH was added. The enzyme activity was calculated and the activity expressed in μg
94 of *p*-nitrophenol released g^{-1} soil h⁻¹ [37]. Alkaline phosphatase (ALP) was measured as that of

95 acid phosphatase [29] with an exception of the change in the pH of *p*-nitrophenyl phosphate as
96 alkaline (pH 11.0).

97 **2.3.4 Fluorescein diacetate (FDA)**

98 FDA hydrolysis was carried out with 2 g of moist soil taken from the experimental setup and the
99 activity measured by spectrophotometry at 490 nm after incubation for 20 min at 30°C,
100 according to the method described by [25]. The FDA hydrolysis rate was expressed as μg
101 fluorescein released g^{-1} soil h^{-1} .

102 **2.4 Statistical analysis**

103 Statistically significant differences between the treatments were analyzed using analysis of
104 variance (ANOVA) and Duncan's Multiple Range Test (DMRT) at 5% level of significance. The
105 principal component analysis (PCA) and Eigenvalues are performed in XLSTAT version
106 2010.5.05 (XLSTAT, 2010).

107 **3. Results and Discussion**

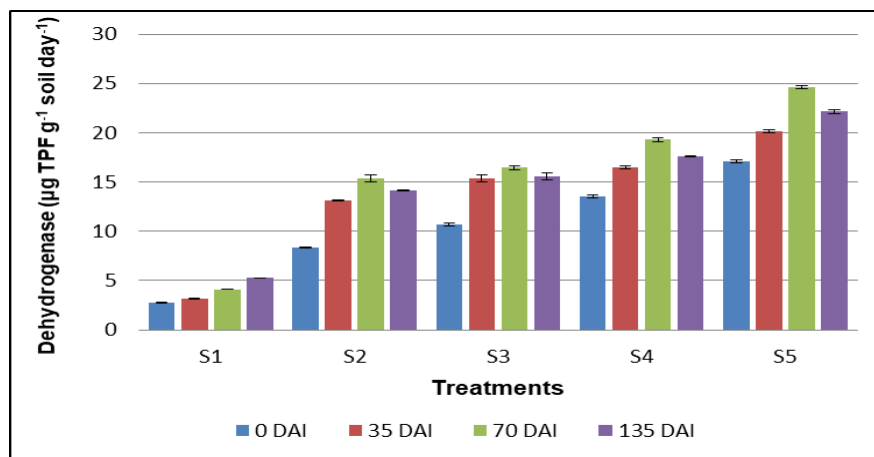
108 Soil enzymes are a crucial factor influencing ecosystem function and are used as biological
109 indicators for assessing the overall soil functions. In the process of the N cycle, nitrification,
110 conversion of ammonia to nitrite and then to nitrate, are a classical two-step reaction. Also, N
111 mineralization is a crucial step for plant N uptake. In order to hasten the process, several N
112 amendments become an integral part of crop management practices. However, the
113 augmentation of these N amendments, more specifically inorganic sources in sustaining soil
114 health, is still a debate. Hence the present investigation was aimed to study the temporal
115 dynamics of soil enzymes pertaining to N cycle under *in vitro* condition in a simulated wetland
116 ecosystem for 150 d.

117 **3.1. Dehydrogenase activity**

118 The addition of inorganic N amendments increased soil enzymes. Dehydrogenase (DHA)
119 activity increased over time with N amendments up to 70 DAI and thereafter a steady decline
120 was observed (Fig. 2). The dehydrogenase activity ranged between 2.73 and 24.59 μg TPF g^{-1}
121 soil day^{-1} irrespective of the treatments and maximum activity was observed only on 70 DAI in
122 S5 (aerated rhizosphere soil enriched with 0.5% NH_4Cl) compared to control ($P = .05$). The
123 increase over time of DHA in NH_4Cl amended soil compared to non-amended and NO_2
124 amended soils indicate the availability of NH_4^+ ions in soil solutions. An increase in DHA activity
125 in S4 showed active metabolic reactions catalyzed by soil microbiome producing adenosine
126 triphosphate through oxidation of organic matter [23]. Furthermore, it signifies efficient N
127 assimilation and increased microbial biomass in NH_4Cl amended soil.

128 Oxygen diffusion rate (ODR) is the proximal regulator of soil microbial activities [15]. Decrease
129 of soil water content ($> \text{pF}$) causes an increase in ODR and redox potential [32]. The reduction
130 of dehydrogenase (DHA) activity beyond 70 DAI might be attributed due to increased redox

131 potential caused by loss of soil moisture. The response of DHA activity in the present study is in
 132 line with the findings of [34], that the activity of dehydrogenase in an inorganic fertilized soil at
 133 different stages of rice crop ranged between $12.75 \mu\text{g TPF g}^{-1} \text{ soil day}^{-1}$ and $44.23 \mu\text{g TPF g}^{-1}$
 134 soil day^{-1} . Thus, soil dehydrogenase activity in the treatments showed a decrease with an
 135 increase in incubation time.



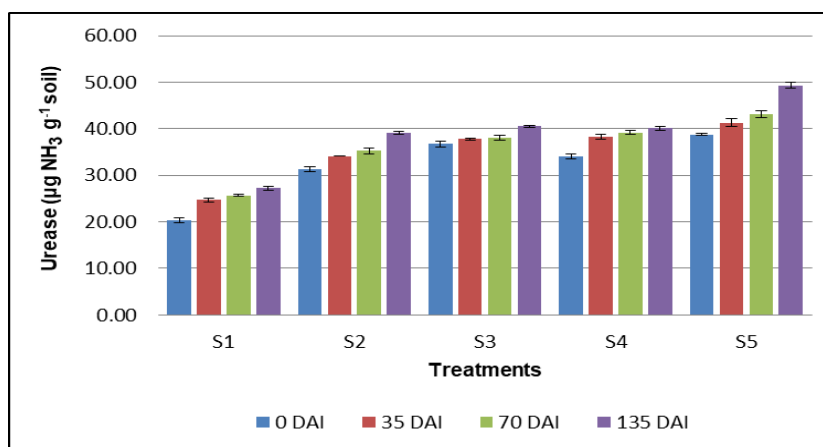
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 137 **Fig. 2. Influence of nitrogen amendment on soil dehydrogenase**

138 Values are mean (\pm standard error) ($n=3$) and within each column, values followed by the same
 139 letters are not significantly different from each other as determined by DMRT ($P \leq .05$). S1 -
 140 Rhizosphere soil; S2 - Rhizosphere soil with aeration; S3 - combined NH_4Cl and KNO_2 enriched
 141 rhizosphere soil with aeration; S4 - KNO_2 enriched rhizosphere soil with aeration; S5 - NH_4Cl
 142 enriched rhizosphere soil with aeration.

143 144 3.2. Urease activity

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 146 The soil urease activity differs with the soil type and organic matter content and also by the
 147 adsorption of the enzyme into the soil organic carbon and mineral particles [33]. Maximum
 148 urease activity was seen on 70 DAI, thereafter decreased when the incubation time prolonged
 149 [17]. Here also, in comparison with other treatments, treatment S5 (NH_4Cl) showed maximum
 150 urease activity of $49.27 \mu\text{g g}^{-1} \text{ soil}$ on 70th day (Fig 3). However, statistical significance was not
 151 observed at $P = .05$, irrespective of the treatments, and DAI. The urease activity depends on the
 152 level of N fertilization [28] and releases $\text{NH}_4\text{-N}$ through urea hydrolysis. It is also essential for
 153 the hydrolysis of amino compounds [24, 30]. The non-significance in urease activity may be due
 154 to the application of urea in the previous season and have a profound influence on microbial
 155 biomass. These results were in concordance with the report of [22], who worked on the
 156 influence of the high quantity of ammonia on the activity of urease. An increase in the
 157 temperature increases the urease activity while the reduction in soil moisture by 10% leads to
 158 reduced urease activity.

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Fig. 3. Influence of nitrogen amendment on soil urease

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Values are mean (\pm standard error) ($n=3$) and within each column, values followed by the same letters are not significantly different from each other as determined by DMRT ($P \leq 0.05$). S1 - Rhizosphere soil; S2 - Rhizosphere soil with aeration; S3 - combined NH_4Cl and KNO_2 enriched rhizosphere soil with aeration; S4 - KNO_2 enriched rhizosphere soil with aeration; S5 - NH_4Cl enriched rhizosphere soil with aeration.

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168 3.3. Phosphatase activity

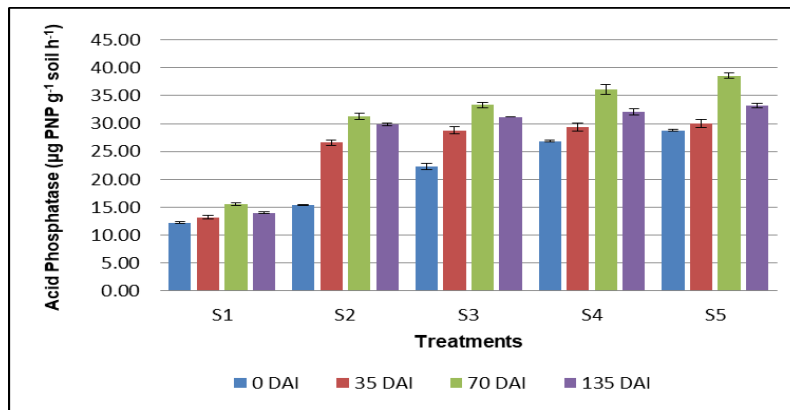
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Phosphorus dynamics in soil depend on pH, N, and organic matter [19, 6]. Similar to DHA and urease, acid monophosphoesterase activity increased up to 70 DAI in all the treatments and after that started declining. The results also coincide with DHA and urease, where maximum acid monophosphoesterase activity was observed in S5 (NH_4Cl) registering $38.57 \mu\text{g PNP released g}^{-1} \text{ soil h}^{-1}$ on the 70 DAI (Fig. 4). However, alkaline phosphatase is more in KNO_2 (S4) amended soils ($53.40 \mu\text{g PNP released g}^{-1} \text{ soil h}^{-1}$) on 70 DAI (Fig. 5) and thereafter declined at a slow rate. The results suggest that N addition exerts a profound influence on soil P availability through changes in microbial metabolism.

177

The increase in acid phosphatase activity in NH_4Cl amended soil might be attributed due to the acidification of soil by ammonium-N. The reduction in soil pH is due to H^+ ions from NH_4^+ . More the NH_4^+ fraction in an amendment will be the acidifying potential and reduction in soil pH [7]. On the contrary, $\text{NO}_2\text{-N}$ could not contribute to soil acidity due to the lack of H^+ ions [27]. Hence the acid phosphatase activity is less in NO_2 amended treatments, whereas alkaline phosphatase activity is more in KNO_2 amended rhizosphere soil.

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185 **Fig. 4. Influence of nitrogen amendment on soil acid phosphatase**

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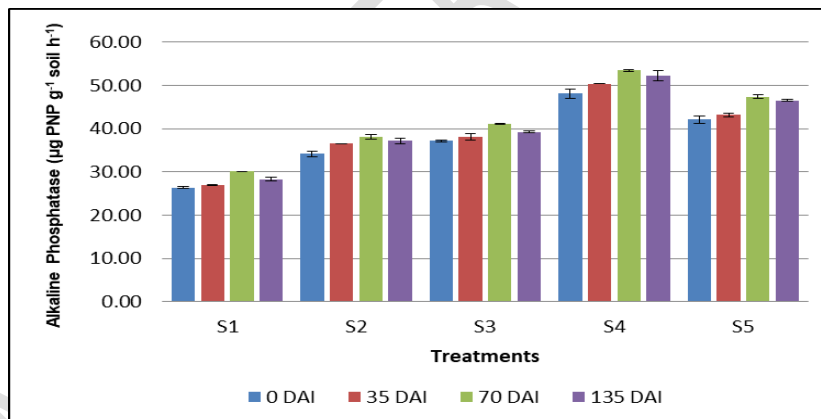
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Values are mean (\pm standard error) ($n=3$) and within each column, values followed by the same letters are not significantly different from each other as determined by DMRT ($P \leq 0.05$). S1 - Rhizosphere soil; S2 - Rhizosphere soil with aeration; S3 - combined NH_4Cl and KNO_2 enriched rhizosphere soil with aeration; S4 - KNO_2 enriched rhizosphere soil with aeration; S5 - NH_4Cl enriched rhizosphere soil with aeration.

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194 **Fig. 5. Influence of nitrogen amendment on soil alkaline phosphatase**

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Values are mean (\pm standard error) ($n=3$) and within each column, values followed by same letters are not significantly different from each other as determined by DMRT ($P \leq 0.05$). S1 - Rhizosphere soil; S2 - Rhizosphere soil with aeration; S3 - combined NH_4Cl and KNO_2 enriched rhizosphere soil with aeration; S4 - KNO_2 enriched rhizosphere soil with aeration; S5 - NH_4Cl enriched rhizosphere soil with aeration.

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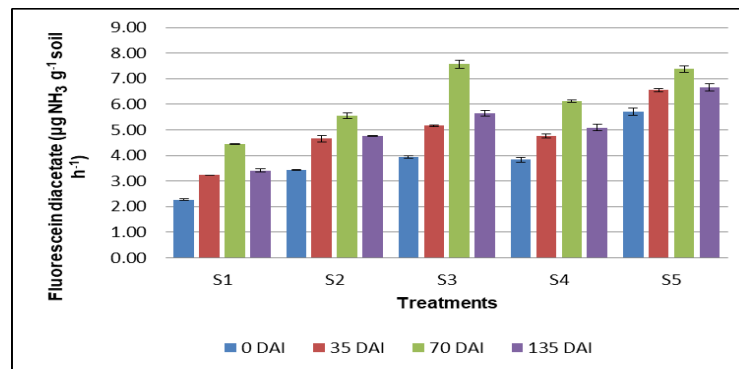
3.4. Fluorescein diacetate activity

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Fluorescein diacetate hydrolysis, an indicator of microbial redox systems represents the detection of microbial oxidative activities in soil [25]. The hydrolysis of the FDA was widespread

203 among the bacteria, fungi, and decomposers. The FDA activity was observed maximum in S3
 204 ($7.57 \mu\text{g}$ fluorescein released g^{-1} soil h^{-1}) with a combined source of $\text{NH}_4\text{-N}$ and $\text{NO}_2\text{-N}$
 205 amended rhizosphere soil when compared to individual compartments (Fig. 6). The results
 206 suggest that both the N sources synergistically contribute towards the soil redox reactions and
 207 indirectly to soil microbial biomass. Accelerated FDA indicates the contribution of several
 208 microbial reactions involved in decompositions of soil organic matter. This in turn indicates the
 209 soil fertility status [20]. Also, the results show concordant with the findings of [26] recorded a
 210 maximum of $19.16 \mu\text{g}$ fluorescein released g^{-1} soil h^{-1} .



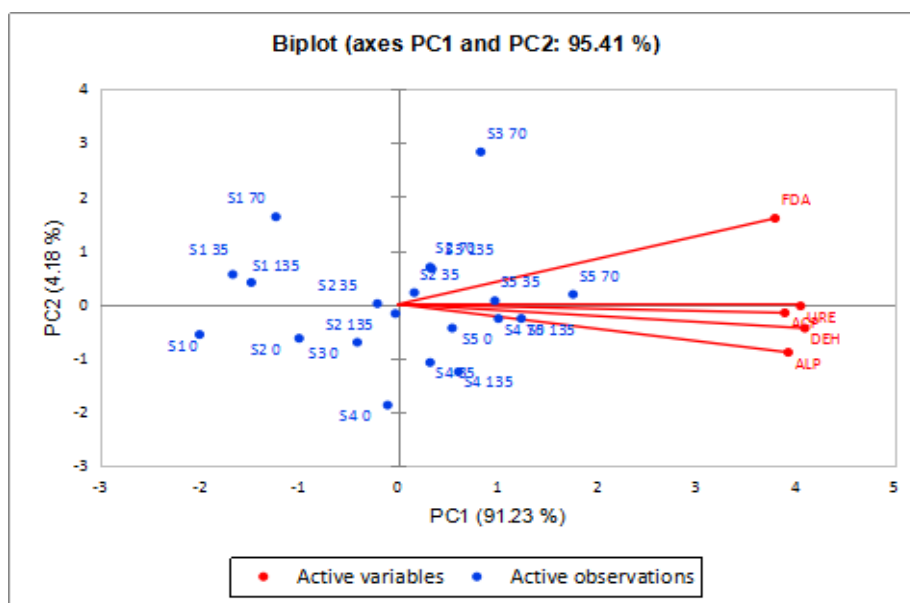
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 212 **Fig. 6. Influence of nitrogen amendment on Fluorescein diacetate**

213 Values are mean (\pm standard error) ($n=3$) and within each column, values followed by the same
 214 letters are not significantly different from each other as determined by DMRT ($P \leq 0.05$). S1 -
 215 Rhizosphere soil; S2 - Rhizosphere soil with aeration; S3 - combined NH_4Cl and KNO_2 enriched
 216 rhizosphere soil with aeration; S4 - KNO_2 enriched rhizosphere soil with aeration; S5 - NH_4Cl
 217 enriched rhizosphere soil with aeration.

218

219 3.5. Principal Component analysis

220 Principal component analysis (PCA) of changes in soil enzyme activities explained 91.23 and
 221 4.18 % variance for PC1 and PC2, respectively (Fig 7). The PC with higher Eigen values >1 and
 222 the variables which had positive factor loading (FDA) were considered as the best
 223 representative of soil enzymes. However, the cumulative variance was 95.41%. In PC1, the
 224 other variables like DHA, URE, ACP, and ALP showing correlation with one another were also
 225 retained for soil quality indexing [2].



226

227 **Fig. 7. Principal component analysis showing the relationship between the soil enzymes**
 228 **in different N amended soil.**

229 DEH-Dehydrogenase, URE-Urease, ACP-Acid Phosphatase, ALP-Alkaline Phosphatase,
 230 FDA-Fluorescein diacetate.

231 4. CONCLUSION

232 The soil enzyme activities responded to **different N amendments** revealed that ammoniacal N
 233 ($\text{NH}_4\text{-N}$) contributed for efficient soil system functioning whereas, combined sources $\text{NH}_4\text{-N}$ and
 234 $\text{NO}_2\text{-N}$ facilitates soil redox reactions and indicates richness in microbial biomass. Also, the
 235 study implies that the addition of N amendments hastens the soil microbiological process and
 236 organic matter decompositions. Hence soil enzymes can be considered as biological indicators
 237 for assessing soil health.

238 COMPETING INTERESTS

239 Authors have declared that no competing interests exist.

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UNDER PEER REVIEW