Original Research Article

Isatin analogues as anti-bacterial agents; QSAR, Docking

Abstract:

Computational chemistry is unique method in drug discovery which reduce cost. In this study 86 molecules containing isatin core were subjected to quantitative structureactivity relationship analysis and docking study to find the structure requirements for ligand binding. The structures were sketched and optimized in Hyperchem. The structural invariants used in this study were those obtained from whole molecular structures: by both hyperchem and dragon software (16 types of descriptors). Four chemometrics methods including MLR, FA-MLR, PCR and GA-PLS were employed to make connection between structural parameters and anticancer effects. MLR analysis explained the positive effect of the number of urea derivatives, thio urea, amide, thioamide, hydrazone, thiocarbohydrazone, nBnz with the halogen substitution on 5 position of isatin ring on the antimicrobial activity. It also shows nArCN, nPyridines have negative effects on the antimicrobial activity of studied compound The FA-MLR describes the effect of 3D-MORSE and Galvez Topological charge descriptors and on antimicrobial activity of the studied compounds. The quality of PCRA equation is better than those derived from FA-MLR. A comparison between the different statistical methods employed revealed that GA-PLS represented superior results and it could explain and predict 73% and 68% of variances in the -LogMIC data, respectively. Comparison between QSAR and docking analysis revealed that by decreasing in number of ring and lipophilicity (also Logp) for design of new compounds can have better activity. Substitutions such as urea, thiourea, thiocarbohydrazone, benzhydrazide as on isatin ring, can cause better interaction with receptor.

Keywords: Isatin, QSAR, Docking, antibacterial

Introduction:

FabI was confirmed to be the only enoyl-acyl carrier protein (ACP) reductase required for the synthesis of fatty acids. FabI is required for the elongation of both long-chain saturated and unsaturated fatty acids in *Escherichia coli*. Thus the activity of this enzyme plays a determinant role in completing cycles of fatty acid biosynthesis in *E. coli*. Indole compounds can inhibit FabI [1]. One of the best promising new indole compounds having many interesting activity profiles are isatin and isatin derivatives. The isatin (1*H*-indole-2,3-dione) moiety is responsible for a wide spectrum of biological property such as antibacterial, antifungal, antiviral, anticancer, anticonvulsant, anti-HIV and antiparkinsonian activity in many synthetically versatile molecules [2–9]. Among these properties antibacterial activities (against *Escherichia coli*) of this moiety was of our interest to study the quantitative structure-activity relationships of a series of 86 isatin derivatives reported in literature.

Synthesis and evaluation of biological activity of novel compounds usually time consuming and take large amounts of money. Today the uses of computational methods for designing newly biologically active compounds have opened a new window to modern drug discovery study. Computational methods can accelerate the procedure of discovering new drugs by designing new compounds and predicting potency or activity of them. Quantitative structure activity relationship (QSAR) studies provide pharmaceutical chemists valuable information that is useful for drug design and prediction of drug activity [10-14]. QSAR studies, as one of the most important areas in chemometrics, give information that is useful for molecular design and medicinal chemistry [8-12]. QSAR models are mathematical equations constructing a relationship between chemical scaffold and biological property. These models have another ability, which is providing a deeper knowledge about molecule design.

Linear and nonlinear QSAR models are mathematical equations that display us enough information about the mechanism of biological activity of compounds by constructing a relationship between chemical structures and biological activities. The first step in constructing QSAR models is the proper representation of the structural and physicochemical features of chemical compounds [15-18]. These features named molecular descriptors that represent variation in the structural features property of the molecules by numerical and have high effect on the biological property of compound [19-22]. Molecular descriptors have been classified into different categories such as physiochemical, constitutional, geometrical, topological, and quantum chemical descriptors. Dragon and hyperchem are two famous computational softwares provide us more than 4000 of these descriptors [23,24].

Different QSAR methods including multiple linear regression (MLR), partial least squares combined with genetic algorithm for variable selection (GA-PLS), factor analysis–MLR (FA-MLR), principal component regression analysis (PCR) were used to make connections between structural descriptors and antibacterial activity of studied compounds [25-28]. An important approach of the researchers in modification of the isatin moiety has been to establish a comprehensive structure–activity relationship (SAR), for this class of antibacterial agents (against *Escherichia coli*).

Our research show that our studied series of compounds didn't evaluate for QSAR studies. Our Different QSAR analysis establishes mathematical relationship between biological activities and computable parameters such as chemical, topological, physicochemical, stereochemical or geometrical and so on indices.

The molecular docking studies help us to understand the different interactions between the ligands and enzyme active sites (FabI) in detail and also help to design novel potent structure. Molecular docking simulation technique was also performed on eighty six compounds to show the details molecular binding models for these compounds interacting with the key active site of protein.

2.Methods

2.1. Data set

The biological data used in this study were antimicrobial activity against Ecoli, (in terms of –log MIC), of a set of 86 isatin derivatives [30-35]. At first the outlier data was removed from the data set according to principle component analysis, it was shown in Figure 1. The data set was classified into calibration and prediction set by kenardston algorithm of the 16 prediction molecules from the spaces of the calculated descriptors. The structural features and biological activity of these compounds are listed in Table 1.

[Table 1. near here], [Figure 1 near here]

2.2. Descriptor generation

The structural features of the studied compounds are listed in Table 1. The twodimensional structures of molecules were drawn by Hyperchem 8.0 software (Hypercube Inc.) to calculate whole molecular structure-based descriptors. The final geometries were obtained with semi-empirical AM1 calculations in Hyperchem program. The molecular structures were optimized using the Polak-Ribiere algorithm until the root mean square gradient was 0.01 kcal mol⁻¹ [23]. Some physicochemical parameters including molecular volume (V), molecular surface area (SA), hydrophobicity (Log P), hydration energy (HE) and molecular polarizability (MP) were calculated using Hyperchem Software. In order to calculate some molecular descriptors including topological, constitutional and functional group descriptors the optimized molecules were transferred into the Dragon package, developed by the Milano chemometrics and QSAR Group [24]. The calculated descriptors from whole molecular structures are briefly described in Table 2.

[Table 2. near here]

3.2. Data screening & model building

The selected descriptors from each class and the experimental data were analyzed by the stepwise regression SPSS (version 22.0) software. The calculated descriptors were collected in a data matrix whose number of rows and columns were the number of molecules and descriptors, respectively. Multiple linear regressions (MLR) and partial

least squares (PLS) were used to derive the QSAR equations and feature selection was performed by the use of genetic algorithm (GA). MLR with factor analysis as the data pre-processing step for variable selection (FA-MLR) and principal component regression analysis (PCRA) methods were also used to explore the QSAR equations.

The resulted models were validated by leave-one out cross-validation procedure (using MATLAB software) to check their predictability and robustness.

A key step in QSAR modeling is evaluating model's stability and prediction ability. We used cross-validation and external test set for these proposes. Crossvalidation has different variants such as leave-one-out (LOO), leave-group-out (LGO) and v-fold. It was shown previously that LOO can leads to chance and overfitted models whereas LGO is more sensitive to chance variables [36]. Therefore, we used LGO for model-validation utilizing correlation coefficient and root mean square error of cross-validation (*q2* and *RMSECV*, respectively) as scoring function. In addition, an external test set composed of 6 molecules was also used. The molecules in this set did not have contribution in the model step and thus their predicted values can give a final prediction power of the models as measured by correlation coefficient, root mean square errors of prediction, relative error of prediction (R^2_P , *RMSEP* and *REP*, respectively).

The PLS regression method used in this study was the NIPALS-based algorithm existed in the chemometrics toolbox of MATLAB software (version 12 Math work Inc.). Leave-one-out cross-validation procedure was used to obtain the optimum number of factors based on the Haaland and Thomas F-ratio criterion [37].

3. Results and discussion

3.1. MLR analysis

In the first step, separate stepwise selection-based MLR analyses were performed using different types of descriptors, and then, an MLR equation was obtained utilizing the pool of all calculated descriptors. The resulted QSAR models from different types of descriptors for the compounds (86 molecules as calibration and 16 molecules as prediction sets) are listed in Table 3.

[Table 3. near here]

The equation E1 of Table 3 shows among chemical descriptors, the negative effect of log p of the molecules on the antimicrobial activity. This equation shows the hydrophilic molecules shows better antimicrobila effect. The second equation of Table 3 demonstrated the effect of constitutional descriptors on the activity of these compounds. It also explain the negative effects nCIC (number of rings), and nR10 (number of 10-membered rings) on activity (such as molecule series 20-39, 49, 55, 65 have intermediate activity and inactive compounds).

The effect of topological group counts parameter on antimicrobial activity of the studied compounds has been described by equation E_3 of Table 3. It shows that among topological descriptors spanning tree number (STN) has the negative effect on cytotoxic activity of the compounds.

The equation E_4 of Table 3 was found by using Mol-Walk descriptors (E_4), which explains the negative effect of PIPC09 of studied compounds on the activity of the compounds. The equation E_5 - E_{16} and E_{16} of Table 3 demonstrated the effect of posititive and negative effects of BCUT, Galvz topological Charge indices, 2D autocorrelations, Charge, Burden eigenvalues, RDF, 3D MoRSE, WHIM, GETAWAY and charge descriptors on the anti-cancer activity of these compounds.

The MLR equation of Table 3 obtained from the pool of functional groups descriptors, E_{17} , explained the positive effect of the number of urea derivatives (nCONN), thio urea (nCSNN) (such as molecules of 1-9, 69-86), amide(nCONH2), thioamide (nRCONHR), hydrazone (nC=N-N), thiocarbohydrazone (such as molecules of 1-9) on the antimicrobial activity. It also indicate the positive effect of nBnz (number of benzene rings) with the halogen substitution on 5 position of isatin ring F > Cl > Br (nArX) (molecules series 14-19). This equation also shows nArCN (Aromatic nitrile such as compounds 22, 25, 28, 31, 34, 37, 39), nPyridines (pyridine derivetives) have negative effects on the antimicrobial activity. May be there isn't any electron with drawing group in the receptor site, thus for design of new compounds it's better to don't use these substitution on the backbone of compounds. The negative sign of this group proposed that a decrease in the number of these descriptors resulted in an activity enhancement. This equation, which has a high statistical quality ($R^2 = 0.63$, $O^2 = 0.59$).

The statistical parameters of prediction, listed in Table 4, indicate the suitability of the proposed QSAR model based on MLR analysis of molecular descriptors. The

correlation coefficient of prediction is 0.62, which means that the resulted QSAR model could predict 62% of variances in the antimicrobial activity data. It has root mean square error of 0.21.

[Table 4. near here]

3.2. GA-PLS model

Multicolinearity is a real problem in MLR analysis. This problem in the descriptors is omitted by PLS analysis. In fact, in PLS analysis, the descriptors data matrix is decomposed to orthogonal matrices with an inner relationship between the dependent and independent variables. This modeling method coincides with noisy data better than MLR, because a minimal number of latent variables are used for modeling in PLS. In GA-PLS analysis a variable selection method is used to find the more convenient set of descriptors because redundant variables degrade the performance of PLS analysis, similar to other regression methods. In the present study, GA was used as variable selection method. The data set (n = 86) was divided into two groups: calibration set (n = 70) and prediction set (n = 16). Given 70 calibration samples; cross-validation procedure was used to find the optimum number of latent variables for each PLS model. In this work, in each run of GA-PLS method a large number of acceptable models were created. GA produces a population of acceptable models in each run. In this work, many different GA-PLS runs were conducted using different initial set of populations (50-250) and therefore a large number of acceptable models were created. The most convenient GA-PLS model that resulted in the best fitness contained 8 descriptors including, one constitutional descriptor (nCIC), one 3D MoRSE descriptors (MOR30M) parameter, two WHIM descriptor (P1P, E1U) and four functional descriptors (nRCONHR, nCONN, nArCN, nPyridines). The majority of these descriptors are functional indices All of them being those obtained by different MLR-based QSAR models. The PLS estimate of the regression coefficients are shown in Figure 2.

This model not only has a high cross-validation statistics, but also represents a high ability for modeling external test samples. It could explain and predict about 75% of variances in the antimicrobial activity (against Ecoli) of the studied molecules. There is a close agreement between the experimental and predicted values of antimicrobial activity data.

To measure the significance of the 8 selected PLS descriptors in the protein tyrosine kinase inhibitory activity; In order to investigate the relative importance of the variable appeared in the final model obtained by GA-PLS method, variable important in projection (VIP) was employed [38]. VIP values reflect the importance of terms in PLS model. According to Erikson *et al.* X-variables (predictor variables) could be classified according to their relevance in explaining y (predicted variable), so that VIP > 1.0 and VIP < 0.8 mean highly or less influential, respectively, and 0.8 < VIP < 1.0 means moderately influential. The VIP analysis of PLS equation is shown in Figure 3. As it is observed, nRCONHR, P1P and E1U indices represent the most significant contribution in the resulted QSAR model. In addition, parameters such as nArCN and MOR30M have been found to be moderately influential parameters.

[Figure 2. Near here], [Figure 3. Near here]

3.3. FA-MLR and PCRA

FA-MLR was performed on the dataset. Factor analysis (FA) was used to reduce the number of variables and to detect structure in the relationships between them. This data-processing step is applied to identify the important predictor variables and to avoid collinearities among them [39]. Principle component regression analysis, PCRA, was tried for the dataset along with FA-MLR. With PCRA collinearities among **X** variables are not a disturbing factor and the number of variables included in the analysis may exceed the number of observations [40]. In this method, factor scores, as obtained from FA, are used as the predictor variables [39]. In PCRA, all descriptors are assumed to be important while the aim of factor analysis is to identify relevant descriptors.

Table 5 shows the nine factor loadings of the variables (after VARIMAX rotation) for the compounds tested for cytotoxic activity. As it is observed, about 81.2% of variances in the original data matrix could be explained by the selected nine factors.

Based on the procedure explained in the experimental section, the following twoparametric equation was derived (Table 6).

Y= 5.766(±0.547) -0.031(±0.012) MOR30m -2.336(±0.721) JGI5 $R^2 = 0.64 \ S.E = 0.24 \ F = 14.69 \ Q^2 = 0.59 \ RMScv = 0.12$ This equation could explain about 59% of the variance and predict 64% of the variance in pMIC data. It has a root mean square error of 0.12. This equation describes the effect of 3D-MORSE and Galvez Topological charge descriptors (MOR30m and JGI5) and on antimicrobial activity of the studied compounds.

When factor scores were used as the predictor parameters in a multiple regression equation using forward selection method (PCRA), the following equation was obtained (Table 7):

 $Y=4.285(\pm 0.57)-0.258(\pm 0.031)F1-.189(\pm 0.021)F7+0.124(\pm 0.027)F6+0.123(0.057)F8$ R²=0.73 S.E. = 0.34 F = 15.54 Q² = 0.68 RMScv = 0.15

This equation could explain and predict 68% and 73% of the variances in pMIC data, respectively. The root mean square error of PCRA analysis was 0.15. Since factor scores are used instead of selected descriptors, and any factor-score contains information from different descriptors, loss of information is thus avoided and the quality of PCRA equation is better than those derived from FA-MLR. Whilst the data of this analysis show acceptable prediction, we see that the predicted values of some molecules are near to each other.

As it is observed from Table 5, in the case of each factor, the loading values for some descriptors are much higher than those of the others. These high values for each factor indicate that this factor contains higher information about which descriptors. It should be noted that all factors have information from all descriptors but the contribution of descriptor in different factors are not equal. For example, factors 1 and 2 have higher Constitutional, Charge, WHIM, Atom-center, Connectivity, Functional, MORSE and GETAWAY whereas information about RDF, MORSE, burden eigenvalues 2Dautocorrelations and functional descriptors are highly incorporated in factor 3 and 4. Factor score 5, 6, 7 and 8 signify the importance of functional chemical and Atom-center descriptors.

[Table 5 near here], [Table 6 near here], [Table 7 near here]

3.4. Robustness and applicability domain of the models

Leverage is one of standard methods for this purpose. Warning leverage (h^*) is another criterion for interpretation of the results. The warning leverage is, generally, fixed at 3k/n, where *n* is the number of training compounds and *k* is the number of model parameters. A leverage greater than warning leverage h^* means that the predicted response is the result of substantial extrapolation of the model and therefore may not be reliable [41]. The calculated leverage values of the test set samples for different models and the warning leverage, as the threshold value for accepted prediction, are listed in Table 8. As seen, the leverages of all test samples are lower than h^* for all models. This means that all predicted values are acceptable.

[Table.8 near here]

3.5. Molecular Docking Studies

The docking study was performed using the AutoDock 4.2. All the eighty six isatin derivatives were docked into the active site of the enzymes FabI (PDB:11x6). All the docking protocols were done on validated structures, with RMSD values below 2 Å. The conformation with the lowest ones was considered as the best docking result. Docking binding energies of these active compounds were summarized in Table 9.

Docking analysis showed that Compounds 1-9 with thiocarbazone moiety, was good inhibitor for FabI, because of good interaction between enzyme and cofactor. With suitable orientation of thiocarbazone group, hydrogen and hydrophobic bounds can occur. An electron rich group such as NH₂ substitution on phenyl ring that increase electron charge can create better interaction with receptor and has low binding energy. Also compounds have benzhydrazide substitution hydrogen binding interaction between tyrosine 156 and benzhydrazide. Halogen and methyl substitution on isatin ring of this series (1-9) can cause better interaction with receptor. Benzyl amide, methyl and halogen groups on isatin ring of compound 10-19 showed good docking score. But Halogen and methyl groups at C-5 isatin ring of series (40-43) can cause bad interaction with receptor. Urea moiety can show good interaction between tyrosine 156 and coenzyme. Our results indicated Benzamide group show good interaction but OH group has bad interaction with receptors. Compounds 47 and 48 with isonicotine amide group on isatin ring showed good interaction with receptor. The interaction modes of 2, 14 and 27, 40 those with the best docking score are shown in Figure 4.

[Table 9near here], [Figure 4 near here]

4.Conclusions

Quantitative relationships between molecular structure and antibacterial activity of isatin derivatives were discovered by four chemometrics methods: MLR, GA-PLS, PCR and FA-MLR. MLR analysis explained the positive effect of the number of urea derivatives (nCONN), thio urea (nCSNN), amide (nCONH2), thioamide (nRCONHR), hydrazone (nC=N-N), thiocarbohydrazone on the antimicrobial activity. It also indicate the positive effect of nBnz (number of benzene rings) with the halogen substitution on 5 position of isatin ring F > Cl > Br (nArX). This equation also shows nArCN, nPyridines (pyridine derivetives) have negative effects on the antimicrobial activity of studied compound. The FA-MLR describes the effect of 3D-MORSE and Galvez Topological charge descriptors (MOR30m and JGI5) and on antimicrobial activity of the studied compounds. The quality of PCRA equation is better than those derived from FA-MLR. Whilst the data of this analysis show acceptable prediction, we see that the predicted values of some molecules are near to each other. Factors 1 and 2 have higher Constitutional, Charge, WHIM, Atom-center, Connectivity, Functional, MORSE and GETAWAY whereas information about RDF, MORSE, burden eigenvalues 2Dautocorrelations and functional descriptors are highly incorporated in factor 3 and 4. Factor score 5, 6, 7 and 8 signify the importance of functional chemical and Atom-center descriptors. A comparison between the different statistical methods employed revealed that GA-PLS represented superior results and it could explain and predict 73% and 68% of variances in the -LogMIC data, respectively. Comparison between QSAR and docking analysis revealed that by decreasing in number of ring and lipophilicity (also logp) for design of new compounds can have better activity. Substitutions such as urea, thiourea, thiocarbohydrazone, benzhydrazide as on isatin ring, can cause better interaction with receptor.

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Table1 1. Chemical structure of isatin derivatives used in this study



Compound	R ₁	R ₂	R ₃	PMIC
1		Н	Н	
				4.00
2		Н	Н	
	H H			4.60

_						
	3**		Н	Н		
					2.70	
	1	S CH	Ц	п	3.70	
	4		п	п		
					4.00	
	5		Н	Н		
		CI			4.60	
	6		Н	Н	4.00	
	-	ОСН₃				
	7**		TT	тт	3.70	
	/**	ОСН3	Н	Н		1
		`OCH₃			4.60	
	8		CH3	Н		
					4.00	
	9		CH ₂	H	4.00	
	,	— 🦑 У—сі	CII,		1.50	
	*		CH	н		
		°OCH₃				
	*	он	Н	CH3		
	*		Н	CH3		
	*		Н	CH3		
		*outlier da	ita			
		**test se	t .			
			L			
	Y					
	<i>F</i>					



Compound	R	PMIC
14	Н	
		5.18
15	2-F	
		6.74
16	3-Cl	
		5.95
17	3-CH3	
		5.31
18	2-Br	5.61
19**	3-F	2.01
-		6.17

thethe



Compound	R	R'	PMIC
20	Н	Н	
			3.60
21	Н	4-CH3	
			3.60
22	Н	4-CN	()
			3.70
23	4-OCH3	Н	
		Y	3.60
24	4-OCH3	4-CH3	
			3.70
25	4-OCH3	4-CN	
			3.82
26	4-C1	Н	
			3.60
27	4-Cl	4-CH3	
			3.70
28	4-C1	4-CN	
			3.60
29	4-CH3	Н	
			3.70
30	4-CH3	4-CH3	
			3.60
31	4-CH3	4-CN	
			3.70
32	4-Br	Н	
			3.30
33	4-Br	4-CH3	
			3.60
34	4-Br	4-CN	
			3.70
35	4-F	Н	
			3.82
36	4-F	4-CH3	3.82

37	4-F	4-CN	
			4.10
38	2-Cl,4-Cl	Н	
			3.60
39**	2-Cl,4-Cl	4-CN	3.82



	Compound	R	Х	PMIC
	40	O II	Н	
		λ.↓		
				4.00
	/1**	0	F	4.90
	71	Ŭ	T	
		SN NH		
				4.90
	42	0	Cl	
		\sim \downarrow		
		N´ NH		
		\sim		
	42	0	CIII2	4.60
	43		СПЭ	
		SN NH		
\checkmark	\mathbf{Y}			
				4.60
	44	0	Н	
		د ال		
	15**	HN NH ₂	Б	4.90
	45.	, ∥	Г	
		SHN NH2		4.30
	46	0	F	
		S N		
		Ĥ []		
				4.90

47	SHN N	F	4.60
48	SHN N	Cl	4.90

UNDER PHILIP



Compound	R	R1	R2	PMIC
49	H	1-naphthyl		
		.r. J		3.78
50	Н	4-Chloro phenyl		
				4.92
51	Н	4-Bromo phenyl		
52	п	4 Mathul phonul		3.78
32	п	4-Methyl phenyl		2.04
53	Н	Phenyl hydrazino		3.94
				3.88
54	Н	Thiosemicarbazino		
				4.17
55	CH3	1-Naphthyl		
56	CU2	4 Chloro nhonyl		3.85
56	CH3	4-Chloro phenyl		2.96
57	NO2	1-Naphthyl		3.86
	$\langle \rangle$	1 5		4 39
58	NO2	4-Bromo phenyl		
				4.18
59	NO2	4-Chloro phenyl		
60	CU2	4 Mathawy nhanyl		3.78
00	СПЭ	4-Methoxy phenyi		2.50
61	СН3	Thiosenicarbazino		3.58
		-		4 40
62	Cl	4-chloro phenyl		
				4.01
63	Cl	4-Methyl phenyl	Н	
				4.85

64	Cl	Thiosemicarbazino	Н	
				4.00
65	Br	1-Naphtyl	Н	
				3.78
66	Br	4-Methoxy phenyl	Н	
				3.79
67	Н	4-Methyl phenyl	CH2-	
			N(C6H5)2	3.89
68	Cl	4-Bromo phenyl	CH2-	
			N(C6H5)2	3.79
*	Br	4-Methoxy phenyl	CH2-	
			N(C6H5)2	

*outlier data

** test set



Compound	R1	R2	PMIC
*	Н	Н	
*	Н	5'-NO2	
69**	Н	5'-Cl,7'-NO2	4 60
70	Н	5'-Br,7'-NO2	4.60
71**	Н	5',7'-diBr	4.00
72**	H	5'-F	4.30
73**	н	5'-Cl	4.30
74	Ĥ	5'-Br	4.30
75**	Н	5'-1	4.30
76**	Н	5'-Me	4.30
77	Н	7'-Me	4.60
78	Н	5'-iPr	4.60
79	Me	Н	4.30
80	Et	Н	4.90
81	n-Pr	Н	4.90
82**	n-Bu	Н	4.90
			4.90

83**	i-Bu	Н	
			4.90
84	Allyl	Н	
			4.00
85	Bn	Н	
			4.00
86	Phen	Н	
			4.00

*outlier data/ **test set

WILL BURNER

Descriptor type	Molecular Description
Chemical	LogP (Octanol-water partition coefficient), Hydration Energy (HE), Polarizability (Pol), Molar refractivity (MR), Molecular volume (V), Molecular surface area (SA).
Constitutional	mean atomic vander Waals volume (MV), no. of atoms, no. of non-H atoms, no. of bonds, no. of heteroatoms, no. of multiple bonds (nBM), no. of aromatic bonds, no. of functional groups (hydroxyl, amine, aldehyde, carbonyl, nitro, nitroso, etc.), no. of rings, no. of circuits, no of H-bond donors, no of H-bond acceptors, no. of Nitrogen atoms (NN), chemical composition, sum of Kier-Hall electrotopological states (Ss), mean atomic polarizability (Mp), number of rotable bonds (RBN), mean atomic Sanderson electronegativity (Me), number of Chlorine atoms (NCl), number of 9-membered rings (NR09), etc.
Topological	Molecular size index, molecular connectivity indices (X1A, X4A, X2v, X1Av, X2Av, X3Av, X4Av), information content index (IC), Sum of topological distances between FF (T(FF)), Ratio of multiple path count to path counts (PCR), Mean information content vertex degree magnitude (IVDM), Eigenvalue sum of Z weighted distance matrix (SEigZ), reciprocal hyper-detour index (Rww), Eigenvalue coefficient sum from adjacency matrix (VEA1), radial centric information index, 2D petijean shape index (PJI2), mean information content on the distance degree equality (IDDE), structural information content (neighborhood symmetry of 3-order) (SIC3), Randic-type eigenvector-based index from adjacency matrix (VRA1), sum of topological distances between OO(T(OO)), etc.
Geometrical	3D-Balaban index (J3D), span R (SPAN), length-to-breadth ratio by WHIM (L/BW), sum of geometrical distances between NN ($G(NN)$), sum of geometrical distances between NO ($G(NO)$), sum of geometrical distances between OO ($G(OO)$), ect.
Walk-Mol	molecular walk count of order 08 (MWC08), self-returning walk count of order 05 (SRW05), total walk count (TWC), etc.
Burden matrix	highest eigenvalue n. 1 of Burden matrix / weighted by atomic masses (BEHM1), highest eigenvalue n. 7 of Burden matrix / weighted by atomic masses (BEHM7), lowest eigenvalue n. 1 of Burden matrix / weighted by atomic masses (BELM1), highest eigenvalue n. 1 of Burden matrix / weighted by atomic van der Waals volumes (BELV1), highest eigenvalue n. 2 of Burden matrix / weighted by atomic Sanderson electronegativities (BEHE2), etc.
Galvez	topological charge index of order 1 (GGI1), topological charge index of order 6 (GGI6),topological charge index of order 7 (GGI7), global topological charge index (JGT), etc.
2D autocorrelation	Broto-Moreau autocorrelation of a topological structure - lag 7 / weighted by atomic Sanderson electronegativities (ATS7E), Moran autocorrelation -lag 4 / weighted by atomic Sanderson electronegativities (MATS4E), Broto-Moreau autocorrelation of a topological structure - lag 3 / weighted by atomic Sanderson electronegativities (ATS3E), Broto-Moreau autocorrelation of a topological structure - lag 3 / weighted by atomic van der Waals volumes (ATS3V), etc.
Charge	maximum positive charge (QPOS), partial charge weighted topological electronic charge (PCWTE), etc.
Aromaticity	Harmonic Oscillator Model of Aromaticity index,RCI;Jug RC index HOMA

Table 2. Brief description of some descriptors used in this study

	aromaticity indices, HOMT; HOMA total (trial), etc.
Randic	DP0;molecular profile, SP0;shape profile; SHP;average shape profile index,
Tunture	etc.
RDF	Radial Distribution Function - 7.0 / unweighted(RDF070U), Radial Distribution
ND1	Function - 13.5 / unweighted(RDF135U), Radial Distribution Function - 1.0 /
	weighted by atomic masses(RDF010M), Radial Distribution Function - 3.0 /
	weighted by atomic masses(RDF030M), Radial Distribution Function - 4.5 /
	weighted by atomic masses(RDF045M).Radial Distribution Function - 12.5 /
	weighted by atomic masses(RFD125M).Radial Distribution Function - 2.0 /
	weighted by atomic van der Waals volumes(RDF020V) Radial Distribution
	Function - 8 5 / weighted by atomic van der Waals volumes(RDF085V) Radial
	Distribution Function - 1.0 / weighted by atomic Sanderson
	electronegativities(RDF010E) etc
2D M-DCE	3D-MoRSE - signal 01 / unweighted (MOR0111)(0111 0211 3211) 3D-
JD-WORSE	MoRSE - signal 01 / weighted by atomic van der Waals volumes (MOR01V)
	01V 02V = 32V) ect
	1st component symmetry directional WHIM index / weighted by atomic
WHIM	nolarizabilities (G1P) 2st component symmetry directional WHIM index /
	weighted by atomic electrotonological states (G2S). D total accessibility index /
	weighted by atomic van der Waals volumes (DV) etc
	H autocorrelation of lag $1 / lag 2 / lag 3$ weighted by atomic Sanderson
GEIAWAY	electronegativities (H1E H2E H3E) total information content on the leverage
	equality (ITH) R maximal autocorrelation of lag 3 / lag4 unweighted
	(R3U+ R4U+) R maximal autocorrelation of lag 6 / weighted by atomic masses
	(R6M+) R maximal autocorrelation of lag 5 / weighted by atomic van der
	Waals volumes ($R5V+$) R maximal autocorrelation of lag 1 / lag 4 weighted by
	atomic Sanderson electronegativities (R1F+) R maximal autocorrelation of lag
	$\frac{3}{weighted by atomic polarizabilities (R12+), R maximal autocorrelation of lag$
	number of total secondary C(sn3) (NCS) number of ring tertiary C(sn3)
Functional	(NCRHR) number of secondary C(sp3) (nes), number of tertiary c(sp3)
	(aliphatic) (NNR2) number of N hydrazines (aromatic) (nN-NPH) number of
	nitriles (alignatic) (NCN) number of phenols (NOHPH) number of ethers
	(aromatic) (NRORPH) number of solfures (NRSR) etc
	$CHR3 (C-003) CR4 (C-004) X_{}CR X (C-034) Ar_{-}C(=X)_{-}R (C-039) R_{}$
Atom-Centred	$C(=X)-X/R_{-}C=X/C_{-}040)$ X-CH X (C-042) H attached to
	C1(sn3) / C0(sn2) (H-047) RCO-N < / >N-X=X (N-072) R2S / RS-SR (S-107)
	etc
	X0(connectivity index chi-0) connectivity index chi-1(x1) average
indiana	connectivity index chi-0(XOA)
mulces	Linder (Deleker Linder) ICO (information content inder) TICO (total
information	Undex (Baladan U Index), ICU(Information content index), IICU(total
indices	information content index)
edge adjacency	EEig01x(Eigenvalue 01),EEig01r(Eigenvalue 01 from edge)
indices	
eigenvalue-based	Eig1v(Leading eigenvalue from van der Waals weighted distance
indices	Eigenvalue sum from mass weighted distance matrix),SEigm
	matrixeigenvalue-based indices

Equation	Descriptors	(+) effect	(-) effect	R^2	F	Q^2	SE
E1	chemical		Logp	0.72	5.3	0.67	0.21
E2	constitutional		NCIC,nR10	0.58	18.52	0.52	0.32
E3	topological		STN	0.214	18.53	0.17	0.43
E4	Walk and path counts		PIPC09	0.283	26.81	0.21	0.53
E5	Connectivity indices	X4A		0.63	18.99	0.58	0.34
E6	Information indices	BIC5		0.38	19.685	0.32	0.36
E7	2D autocorrelation	GATS1M	MATS3E	0.64	10.417	0.58	0.17
E8	Edge adjacency indices		EEIG03X	0.231	20.45	0.18	0.55
Е9	Burden eigenvalues		BEHm1	0.63	28.562	0.58	0.28
E10	Topological charge indices		JGI5	0.68	7.50	0.54	0.34
E11	Eigenvalue- based indices		LP1	0.25	23.358	0.17	0.54
E12	Geometrical descriptors	QY	G(NF),DISP V	0.46	8.704	0.39	0.41
E13	RDF descriptors	RDF030M	RDF020M	0.65	10.05	0.59	0.23
E14	3D MoRSE descriptors		MOR30M,M OR24U	0.69	17.67	0.62	0.38
E15	WHIM descriptors	E1U,G1M	E3U,P1P	0.61	15.25	0.56	0.46
E16	GETAWAY descriptors	R4M	R2V,HOU,HT M	0.59	10.88	0.53	0.39
E17	Fuctional group counts	nBnz, nCONN, nRCONHR, nArX, nC=N-N	nArCN, nPyridines	0.63	15.29	0.59	0.34
E18	Atom-centred fragments	C-039,C-034		0.56	13.51	0.49	0.24
E19	Charge descriptors	QMEAN, QPOS		0.58	13.52	0.52	0.54

Table 3. The results of MLR analysis with different types of descriptors.

Table 4.	Statistical	parameters	for testin	g prediction	ability	of the	MLR,	GA-PLS	,
PCR, and	IFA-MLR	models							

Model	\mathbf{R}^2	R ² LOOCV	RMSEcv	R ² p	RMSEp
MLR	0.67	0.62	0.21	0.74	0.18
GA-PLS	0.81	0.75	0.19	0.87	0.23
PCR	0.73	0.68	0.16	0.78	0.22
FA-MLR	0.64	0.59	0.15	0.70	0.14

 R^2 : Regression Coefficient for Calibration set R^2_{LOOCV} : Regression Coefficient for Leave One Out Cross Validation $RMSE_{cv}$: Root Mean Square Error of cross validation $RMSE_p$: Regression Coefficient for prediction set $RMSE_p$: Root Mean Square Error of prediction set

					Component					
	F1	F2	F3	F4	F5	F6	F7	F8	F9	extraction
LOGP	.190	.152	.038	175	.002	.091	.724	083	123	.646
NCIC	.897	.188	017	184	060	107	.013	.137	030	.909
NR10	.774	.073	091	021	.210	.398	.064	.254	063	.887
X4A	655	587	004	101	173	085	.006	.283	.105	.912
BIC5	304	314	.059	.025	444	.461	225	.373	035	.795
MATS3E	.456	.541	085	100	.145	301	.304	.094	286	.812
GATS1M	293	.306	664	.155	050	.184	002	.004	106	.691
BEHM1	006	.183	.851	.034	011	074	017	.001	147	.785
JGI3	.004	125	106	.094	.009	.932	.078	006	017	.910
JGI5	.325	.150	009	.105	.818	.107	040	074	.049	.830
L/Bw	319	787	.151	090	008	197	.129	.042	.059	.813
DISPV	059	107	.286	194	.547	357	.240	047	179	.653
RDF020M	176	.316	.653	.314	214	022	.245	027	144	.783
RDF030M	.019	.893	.188	021	.007	.026	.182	.003	023	.869
RDF025P	.485	.724	104	190	.170	219	.190	.045	078	.927
MOR24U	.230	099	006	719	.283	156	.092	.312	.046	.792
MOR30M	.631	.100	125	.102	282	142	.221	394	.014	.738
MOR32V	002	.245	.067	.757	.268	.076	113	.104	.011	.739
E1U	736	493	028	006	211	136	.002	.148	.011	.872
E3U	.128	543	.089	457	215	- 196	.215	- 259	.278	.803
G1M	073	677	.092	079	148	.237	312	159	.293	.764
P1P	013	- 898	035	- 285	.090	.011	.048	159	065	.930
H0U	.150	.257	- 114	322	.354	- 172	.537	- 259	.226	.767
HTM	020	747	581	092	172	- 120	073	039	015	956
R4M	079	709	444	218	029	062	052	- 032	259	829
R2V	.773	.204	144	154	.031	063	.225	.018	.401	.900
nCb	.728	199	158	150	.180	237	069	010	096	.720
NRCONHR	592	406	.136	.290	107	205	411	.066	.006	.845
NCONN	- 708	2.82	- 109	211	162	125	290	300	- 136	872
NARCN	617	033	028	203	179	036	107	112	- 012	480
nC=N-N	- 687	035	- 107	- 158	158	- 006	139	540	080	851
NArX	140	- 046	652	- 024	229	173	- 350	- 160	260	746
nPyridines	- 062	- 069	- 037	- 007	000	- 009	- 084	- 025	876	784
C-034	.002	037	064	052	151	.010	- 119	.839	051	.752
C-039	- 868	- 138	- 051	082	- 260	- 087	- 092	070	016	870
OPOS	- 367	881	- 057	111	027	- 161	082	029	- 038	962
OMEAN	- 732	348	- 053	222	085	161	- 304	123	048	853
% variance	21.769	19.668	7.805	5.894	5.753	5.520	5.368	5.277	4.153	81.208

Table 5. Numerical values of factor loading numbers 1–9 for descriptors afterVARIMAX rotation

	Unstandardized Coefficients		Standardized Coefficients	t	Sig.	R ²	F	Q^2	SE
	В	Std. Error	Beta						
(Constant)	5.766	.547		10.537	.000	0.64	14.691	0.59	0.24
MOR30m	-0.031	.012	303	-2.619	.000				
JGI5	-2.336	0.721	442	-3.240	.002				

Table 6. The results of FA-MLR analysis with different types of descriptors

	Unstandardized Coefficients		Standardized Coefficients	t	Sig.	R^2	F	Q^2	SE
	В	Std. Error	Beta						
(Constant)	4.285	.057		75.368	.000				
F1	258	.031	409	-4.510	.000	0.73	17.16	0.68	0.34
F7	189	.021	300	-3.305	.001				
F6	.124	.027	.197	2.176	.033				
F8	.123	.057	.196	2.159	.034				

 Table 7. The results of PCR analysis

Molecule .no	MLR	GA-PLS	PCR	FA-MLR
3	0.117418	0.084802	0.026426	0.028202
7	0.058532	0.067963	0.03644	0.034729
10	0.087443	0.157524	0.131804	0.035335
11	0.071099	0.093302	0.092915	0.021066
19	0.054337	0.08314	0.03296	0.037432
39	0.081619	0.077263	0.136844	0.040156
41	0.097168	0.134119	0.13121	0.056011
45	0.158855	0.144921	0.152167	0.036003
69	0.045048	0.101806	0.06149	0.068055
71	0.109807	0.13409	0.023009	0.022631
72	0.102708	0.297308	0.041009	0.060281
73	0.105906	0.198805	0.022111	0.063121
75	0.087529	0.127991	0.018691	0.025659
76	0.04769	0.084609	0.021734	0.045611
82	0.081846	0.058078	0.022526	0.016686
83	0.077447	0.07017	0.016547	0.014426
h*	0.214286	0.342857	0.191429	0.108571
S				

Tabel 8. Leverage (*h*) of the external test set molecules for different models. The last row (h^*) is the warning leverage.

Compounds	ΔG kcal/ mol	Κ _i (μΜ)	Atom of the ligand	Receptor/Coenzy me	Interaction	Distance(A°)
1	-8.46	631.45	C=O (isatine)	Tyr 156	Hydrogen	1.80
			C=O (isatine)	2'-OH	bond	1.85
				(ribose)NAD ⁺	Hydrogen	
					bond	
2	-9.01	248.47	C=O (isatine)	Tyr 156	Hydrogen	1.82
			C=O (isatine)	2'-ОН	bond	1.83
			Isatine	(ribose)NAD ⁺	Hydrogen	
			Indole	NAD' -Pyridine	bond	4
			Indole	Phe 94	P1-P1	\mathbf{A}
				Ald 95	H-Pl U ni	
3	-8.58	512 32	C=O(isatine)	Tyr 156	Hydrogen	1.81
5	-0.50	512.52	C=O (isatine)	2'-OH	hond	1.81
			Isatine	(ribose)NAD ⁺	Hydrogen	1.09
			ibutile	NAD^+ -Pyridine	bond	
				, j	Pi-Pi	
4	-8.86	322.58	C=O (isatine)	Tyr 156	Hydrogen	1.77
			C=O (isatine)	2'-ОН	bond	1.94
			C=S	(ribose)NAD ⁺	Hydrogen	2.66
			Isatine	NH_2 - NAD^+	bond	
			Indole	Pyridine- NAD	Hydrogen	
				Phe 94	bond D: D:	
					P1-P1	
5	8 67	113 00	C=O(isatina)	Tyr 156	Hydrogen	1 78
5	-0.07	443.90	C=O (isatine)	2'-OH	bond	1.78
			Isatine	$(ribose)NAD^+$	Hydrogen	1.09
			Indole	NAD^+ -Pvridine	bond	
				Phe 94	Pi-Pi	
					H-Pi	
6	-8.24	912.17	C=O (isatine)	Tyr 156	Hydrogen	1.85
			C=O (isatine)	2'-ОН	bond	1.88
		\mathbf{A}	Isatine	(ribose)NAD ⁺	Hydrogen	
				NAD' -Pyridine	bond	
7	0.51	577.00	C-O (instinu)	T 1 <i>5 (</i>	P1-P1	1.05
/	-8.51	577.80	C=O (isatine)	1 yr 156 2' OU	Hydrogen	1.95
			C-O (Isatilie)	$(ribose)NAD^+$	Hydrogen	2.09
			isatine	NAD^+ -Pyridine	bond	
				T(TE) T filance	Pi-Pi	
8	-8.46	630.94	C=O (isatine)	Tyr 156	Hydrogen	1.89
			C=O (isatine)	2'-OH	bond	1.95
				(ribose)NAD ⁺	Hydrogen	
					bond	
9	-8.47	614.00	C=O (isatine)	2'-OH	H-Pi	
			NH-thiocarboydrazone	(ribose)NAD ⁺	Hydrogen	1.87
			Isatine	Met 159	bond	2.35
				1 yr 156	Hydrogen	
				Ala 95	uona H_Pi	
14	7 4 4	2 5 7	C-O(bonzobudrozid	Tyr 156	Hydrogen	2 2 2
1 ¹⁴	-7.44	5.52			hand	2.22
			e)	2 -OH-	bona	2.17
	1		C=O(benzohydrazid	ribose(NAD ⁺)	Hydrogen	2.66

Table9. Binding interaction of some studied compounds in active site of enzyme

				e)	Gly 93	bond	
				C=O(isatine)		Hydrogen	
						bond	
15		-7.94	1.52	C=O(benzohydrazid	Tyr 156	Hydrogen	1.88
				e)	2´-OH-	bond	1.52
				C=O(benzohydrazid	ribose(NAD ⁺)	Hydrogen	2.02
				e)	Gly 93	bond	
				C=O(isatine)	-	Hydrogen	
						bond	
16		-7.87	1.66	C=O(benzohydrazid	Tyr 156	Hydrogen	2.19
				e)	2′-OH-	bond	2.03
				C=O(benzohydrazid	ribose(NAD ⁺)	Hydrogen	2.36
				e)	Gly 93	bond	
				C=O(isatine)	Tyr 146	Hydrogen	
				Phenyl		bond	
				,		H-Pi	
17		-7.79	1.95	C=O(benzohydrazid	Tyr 156	Hydrogen	1.97
				e)	2′-OH-	bond	2.12
				C=O(benzohydrazid	ribose(NAD ⁺)	Hydrogen	2.92
				e)	Gly 93	bond	
				C=O(isatine)	Tyr 146	Hydrogen	
				Phenyl		bond	
				,		H-Pi	
18		-7.81	1.88	C=O(benzohydrazid	Tyr 156	Hydrogen	1.98
				e)	2′-OH-	bond	1.63
				C=O(benzohydrazid	$ribose(NAD^{+})$	Hydrogen	3.15
				e)	Gly 93	bond	
				C=O(isatine)	5	Hydrogen	
						bond	
				Y			
19		-7.86	1.72	C=O(benzohydrazid	Tyr 156	Hydrogen	1.79
			\checkmark	e)	, 2′-ОН-	bond	2.23
				C=O(benzohydrazid	$ribose(NAD^{+})$	Hydrogen	2.46
			Y	e)	Glv 93	bond	
				C=O(isatine)	5	Hydrogen	
			7	, <i>,</i>		bond	
	20	-6.57	15.30	Pyrazole	Tyr 156	H-Pi	
) _					
	21	-6.42	19.80	Pyrazole	Tyr 156	H-Pi	
	22	-6.21	28.20				
	23	-6.28	25.03				
	24	-6.30	26.91				
	25	-6.79	10.57	Pyrazole	Tyr 156	H-Pi	
	26	-6.33	27.55	,			
	28	-6 25	26.00				
	29	-6.46	18.25				
	30	-6.03	38 1/				
	50	-0.05	50.14				

	31	-6.37	21.35				
	32	-6.35	22.18				
	33	-6.12	32.63				
	34	-6.30	24.20				
	35						
	36	-6.26	26.17				
	37	-6.65	13.38				
-	38	-6.65	13.38	2-Cl	Leu100	Hydrogen	2.90
	20	6 6 9	12.61	2 (1	Lou100	Hydrogon	2 0 2
	29	-0.06	12.01	2-01	Leuioo	bond	2.65
40		Q // 7	618 68	C=O (isatine)	$NH_{2}(NAD^{+})$	Hydrogen	218
		-0.47	nano	N-H (isatine) C=O (imidazolidine) C=O (imidazolidine)	Phosphate (NAD ⁺) Tyr 156 2'-OH (ribose)NAD ⁺	bond Hydrogen bond Hydrogen bond Hydrogen bond	2.33 1.81 1.83
41		-8.30	828.33 nano	C=O (isatine) C=O (isatine) Isatine	Tyr 156 2'-OH (ribose)NAD ⁺ NAD ⁺ -Pyridine	Hydrogen bond Hydrogen bond Pi-Pi	1.70 1.76
42		-8.08	1.19	C=O (isatine) C=O (isatine) Isatine	Tyr 156 2'-OH (ribose)NAD ⁺ NAD ⁺ -Pyridine	Hydrogen bond Hydrogen bond Pi-Pi	1.81 1.89
43		-7.93	1.53	C=O (isatine) C=O (isatine) Isatine	Tyr 156 2'-OH (ribose)NAD ⁺ NAD ⁺ -Pyridine	Hydrogen bond Hydrogen bond Pi-Pi	1.77 1.94
44		-8.39	709.17 nano	C=O (isatine) C=O (isatine) NH ₂ (urea) NH ₂ (urea)	Tyr 156 2'-OH (ribose)NAD ⁺ Phosphate (NAD ⁺) Phosphate (NAD ⁺)	Hydrogen bond Hydrogen bond Hydrogen bond Hydrogen bond	1.91 1.64 2.81 3.74
46		-8.24	912.17	C=O (isatine) C=O (benzamide) C=O (benzamide)	Gly 93 Tyr 156 2'-OH (ribose)NAD ⁺	Hydrogen bond Hydrogen bond Hydrogen bond	2.44 1.86 2.30
47		-8.33	778.55 nano	C=O (isatine) C=O (isatine) C=O (nicotinamide)	Tyr 156 2'-OH (ribose)NAD ⁺ Gly 93	Hydrogen bond Hydrogen bond	1.81 1.81 2.10

					Hydrogen bond	
48	-8.22	943.86 nano	C=O (isatine) C=O (isatine) C=O (nicotinamide)	Tyr 156 2'-OH (ribose)NAD ⁺ Gly 93	Hydrogen bond Hydrogen bond Hydrogen bond	1.97 1.63 2.80
49	-8.57	519.34	C=O (isatine) C=O (isatine)	Tyr 156 2'-OH (ribose)NAD ⁺	Hydrogen bond Hydrogen bond	1.83 1.68
50	-8.47	618.68	C=O (isatine) C=O (isatine)	Tyr 156 2'-OH (ribose)NAD ⁺	Hydrogen bond Hydrogen bond	2.01 1.76
52	-7.58	2.79 (μM)	C=O (isatine) C=O (isatine)	Tyr 156 2'-OH (ribose)NAD ⁺	Hydrogen bond Hydrogen bond	1.81 1.85
53	-7.80	1.90 (μM)	C=O (isatine) C=O (isatine) Isatine	Tyr 156 2'-OH (ribose)NAD ⁺ NAD ⁺	Hydrogen bond Hydrogen bond Pi-Pi	1.87 2.05

Figure 1. Outlier data by Principle Component Analysis before QSAR analysis



UNDER PHILIP





Figure 3. Plot of variables important in projection (VIP) for the descriptors used in GA-PLS model.





Figure 4. The docked configuration of 2 (A), 14(B), 27(C) and 40 (D) in the binding site of FabI